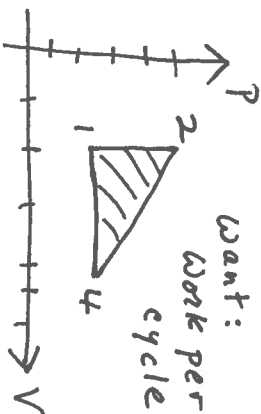


15-heat Engine

①



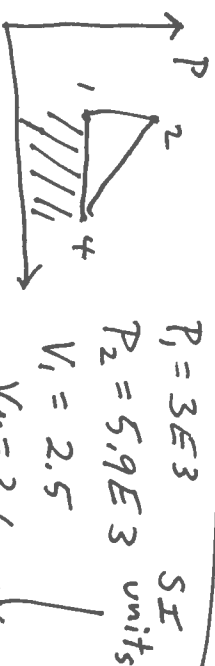
Given: $P_1 = 1.4 \text{ E } 3$ $V_1 = 2.8 \text{ m}^3$
 $P_2 = 2.8 \text{ E } 3$ $V_4 = 5.1 \text{ m}^3$

Work per cycle is area:
 $W = \frac{1}{2} \text{ base} \cdot \text{height}$

base = $5.1 - 2.8$
 height = $(2.8 - 1.4) \cdot 10^3$

Work = $1.61 \text{ E } 3 \text{ J}$

②

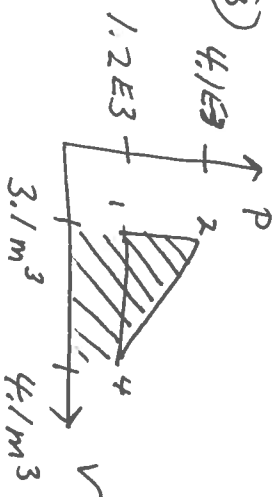


$P_1 = 3 \text{ E } 3$
 $P_2 = 5.9 \text{ E } 3$
 $V_1 = 2.5$
 $V_4 = 3.6$

$W = \text{base} \cdot \text{height}$
 base = $V_4 - V_1 = 3.6 - 2.5$
 height = $P_2 - P_1 = 3 \text{ E } 3$

Work = $3(3.6 - 2.5) \text{ E } 3 = 3.3 \text{ E } 3 \text{ J}$

③



Want: "4"
 Work = $\int P dV$
 = Area under curve.

Area = Rectangle + Triangle

= $P_1(V_4 - V_1) + \frac{1}{2}(P_2 - P_1)(V_4 - V_1)$

= $\left\{ P_1 + \frac{1}{2} P_2 - \frac{1}{2} P_1 \right\} (V_4 - V_1)$

= $\frac{1}{2} (P_1 + P_2) (V_4 - V_1)$

= $\frac{1}{2} (1.2 + 4.1) (\text{E } 3) (4.1 - 3.1) = 3.18 \text{ E } 3 \text{ J}$

④ $PV = nRT \rightarrow T = \frac{PV}{nR}$

$R = \text{gas constant} = 8.314$ SI units (with mols)

$n = 1.4 \text{ mol}$
 $P = 2.2 \text{ E } 3$ Pascals (SI unit)
 $V = 3.3 \text{ m}^3$

$T = \text{Temperature} = \frac{(2.2)(3.3 \text{ E } 3)}{(1.4)(8.314)}$

$T = 624 \text{ Kelvin}$